

**SBI4U Unit 1 Test: Biochemistry**  
**(50 Marks Total)**

Name: \_\_\_\_\_

Signature: \_\_\_\_\_

**Marks obtained:**

Category	Total Marks	Possible Marks
Knowledge/Understanding (K/U)		10
Thinking/Investigation (T/I)		15
Communication (C)		10
Application (A)		15
Total		50
Percentage		

**SECTION 1: Knowledge/Understanding - Multiple Choice (Questions 1-10)***[K/U, 10: 1 each]***Write your section 1 answers here:**

Question	1	2	3	4	5	6	7	8	9	10
Answer										

K/U	T/I	A	C
/10			

**\*\*NOTE: FOR SECTIONS 1 WRITE YOUR ANSWERS IN THE TABLES ON THE FIRST PAGE OF THIS TEST\*\***

**SECTION 1: Knowledge/Understanding - Multiple Choice (Questions 1-10)**

***[K/U, 10: 1 each]***

1. Covalent bonds form when
  - a) two molecules of water share electrons.
  - b) a molecule of water becomes an ion.
  - c) two hydrogen atoms share electrons with one oxygen atom.
  - d) two hydronium ions are attracted to each other.
  
2. The building blocks of most biomolecules contain the element
  - a) carbon.
  - b) calcium.
  - c) nitrogen.
  - d) sodium.
  
3. Which of the following molecules are classified as carbohydrates?
  - a) amino acids
  - b) nucleotides
  - c) fats
  - d) sugars
  
4. Lipids are
  - a) soluble in water.
  - b) made of chains of amino acids.
  - c) linked together with peptide bonds.
  - d) used by the body for storing energy.
  
5. The two types of nucleic acids are
  - a) RNA and ATP.
  - b) DNA and ATP.
  - c) DNA and RNA.
  - d) nucleotides and ATP.

6. An enzyme
  - a) is used up in a reaction.
  - b) raises the activation energy of a reaction.
  - c) bonds with an active site on a substrate molecule.
  - d) lowers the activation energy of a reaction.
  
7. Removing materials from a cell in vesicles is called
  - a) osmosis.
  - b) exocytosis.
  - c) diffusion.
  - d) endocytosis.
  
8. The diffusion of water across a selectively permeable membrane is called
  - a) osmotic pressure.
  - b) pinocytosis.
  - c) osmosis.
  - d) active transport.
  
9. During diffusion, when the concentration of molecules on both sides of a membrane is the same, the molecules will
  - a) move across the membrane to the outside of the cell.
  - b) stop moving across the membrane.
  - c) continue to move across the membrane in both directions.
  - d) move across the membrane to the inside of the cell.
  
10. The part of the enzyme where the substrate binds is called the:
  - a) catalyst
  - b) active site
  - c) inhibitor
  - d) large subunit

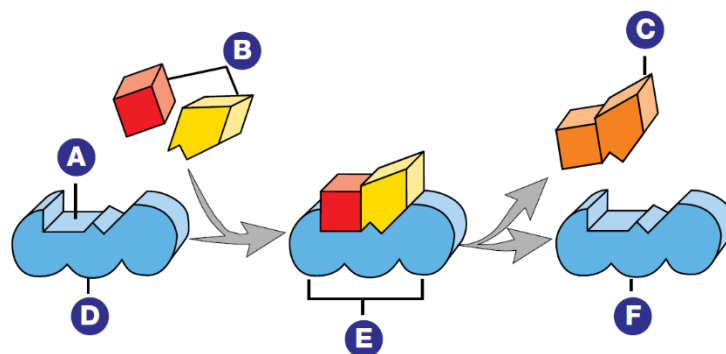
**SECTION 2: Application – Labeling (Questions 11-12)**

**[A, 10: 1 each]**

11.

- a) Label the diagram with the following terms by writing the associated letter on the line:  
[A, 5]

\_\_\_\_\_ Substrates  
\_\_\_\_\_ Enzyme  
\_\_\_\_\_ Active site  
\_\_\_\_\_ Product  
\_\_\_\_\_ Enzyme–substrate complex



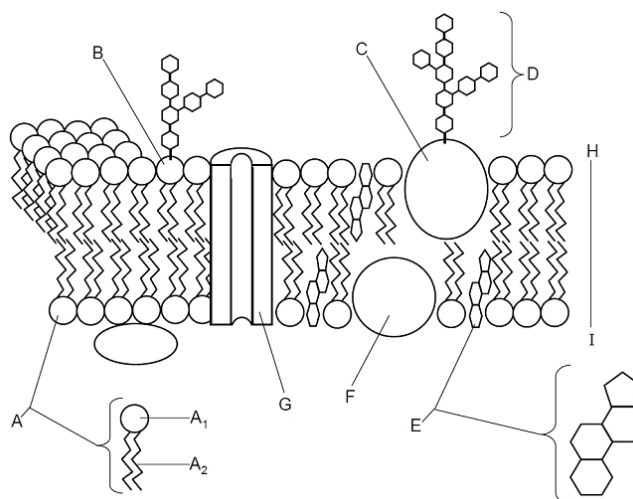
- b) How is enzyme shape important to the enzyme–substrate complex?

[A, 1]

12. Label the diagram with the following terms:

[A, 4]

\_\_\_\_\_ Integral membrane proteins  
\_\_\_\_\_ Peripheral proteins  
\_\_\_\_\_ Carbohydrate molecules  
\_\_\_\_\_ Cholesterol molecules



K/U	T/I	A	C
		/10	

**SECTION 3: Thinking/Investigation, Application & Communication – Short Answer**  
**(Questions 13-18)**

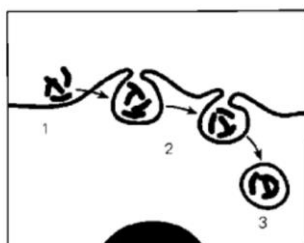
[T/I, 15; A, 5; C, 10]

13. Explain and compare competitive inhibition and noncompetition inhibition.

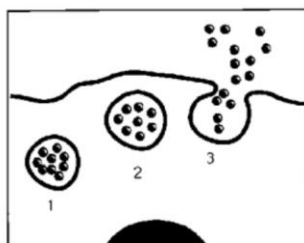
[T/I, 4]

14. Refer to the illustration above. Identify and explain the processes taking place in figure A and figure B.

[A, 5]



**A**




**B**

K/U	T/I	A	C
	/4	/5	

Diagram illustrating the hydrolysis of Lactose:

Lactose (a disaccharide composed of D-Glucose and D-Galactose) is hydrolyzed by the enzyme Lactase, releasing D-Glucose and D-Galactose (monosaccharides) and water ( $H_2O$ ).



The diagram shows the chemical structures of Lactose, D-Galactose, and D-Glucose. Lactose is a disaccharide consisting of a glucose molecule and a galactose molecule linked by a  $\beta$ -1,4-glycosidic bond. The reaction is catalyzed by the enzyme Lactase and involves the addition of water ( $H_2O$ ). The products are D-Galactose and D-Glucose, both monosaccharides in their cyclic forms.

K/U	T/I	A	C
	/11		

17. Use diagrams to show how hydrolysis, condensation reactions, and redox reactions work to form or break bonds in biochemical reactions. *[C, 2 each; 6 total]*

18. Make a generalized diagram to show two amino acids with the bond labeled. *[C, 4]*

K/U	T/I	A	C
			/10